

Order Disorder & Order *by* Disorder



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University
of Windsor

Three questions:

1. What is **order**?

2. What is **disorder**?

3. How can **disorder** lead to **order**?

Three questions:

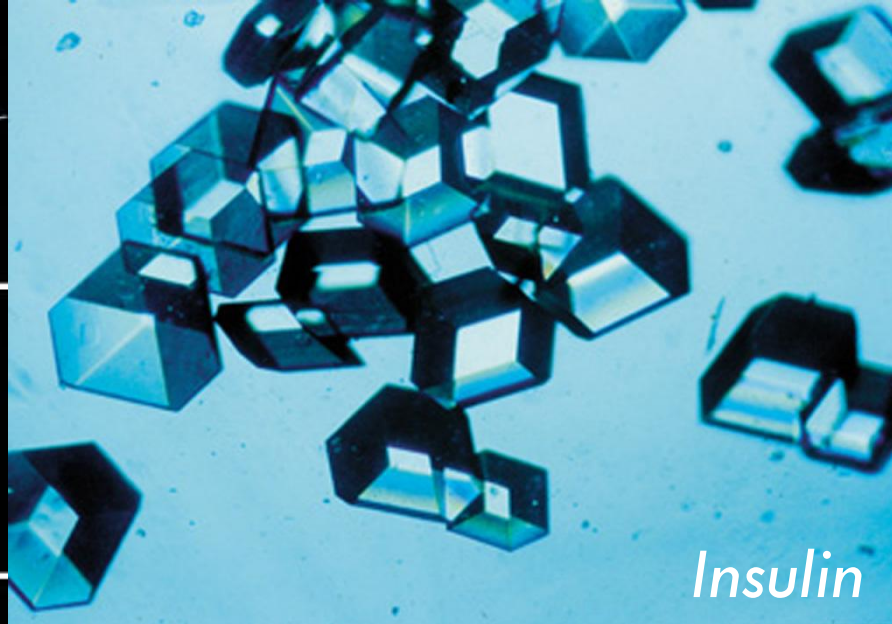
1. What is **order**?

2. What is **disorder**?

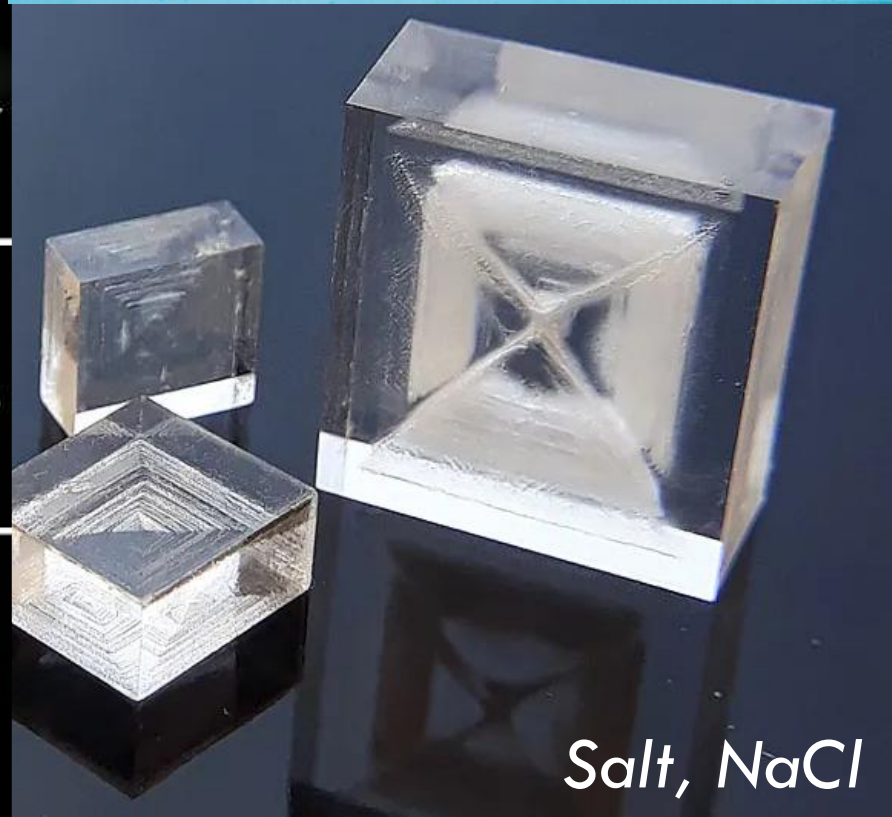
3. How can **disorder** lead to **order**?



Ice



Insulin



Salt, NaCl



Pyrite, FeS_2



"Cave of the Crystals", Mexico

Gypsum, $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$

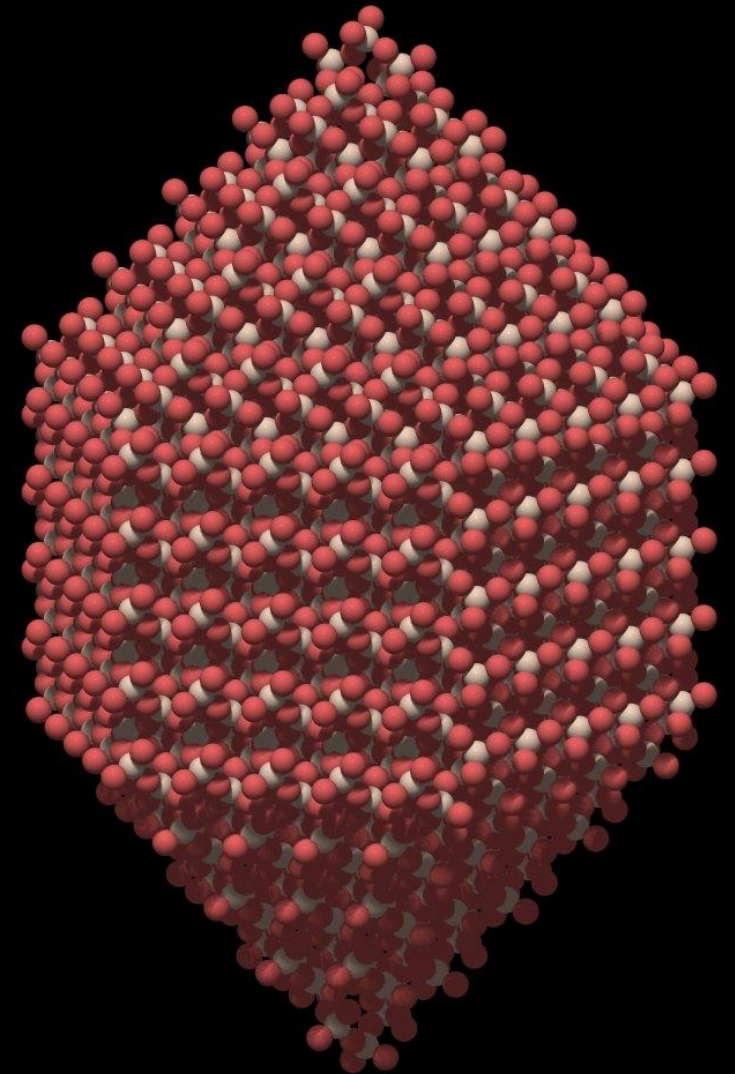
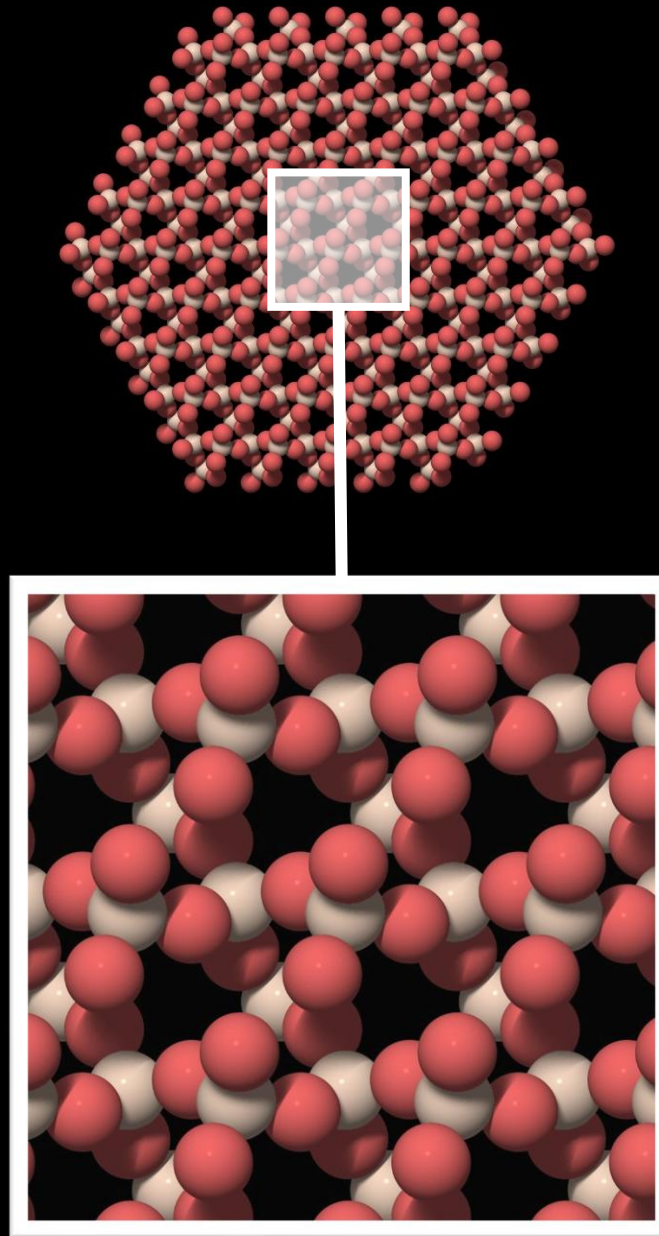
Quartz, SiO_2

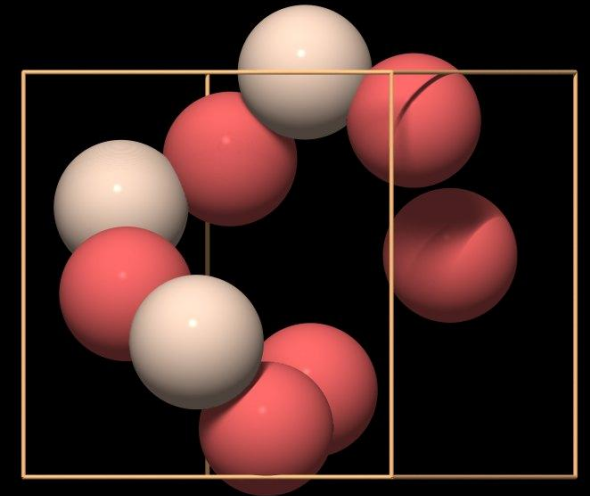
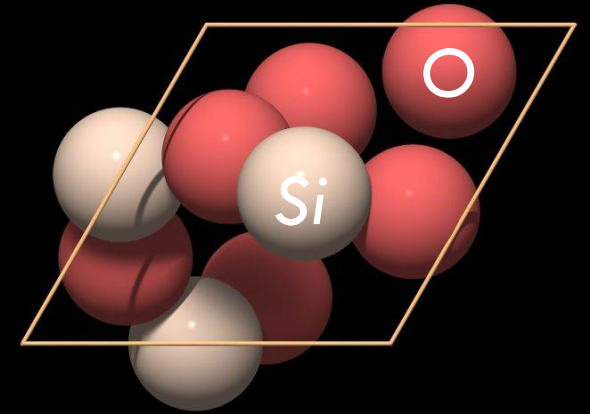
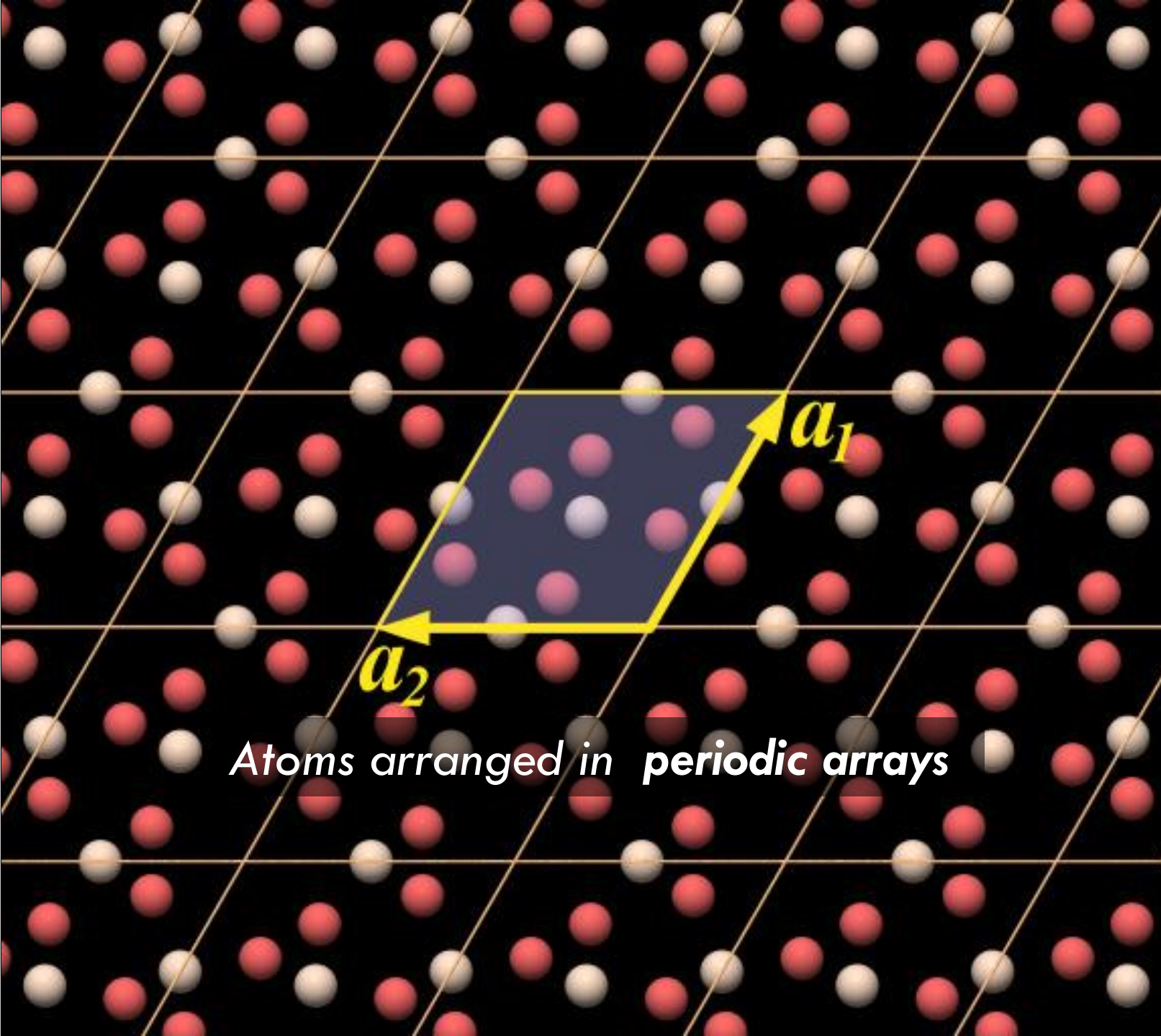


Cut through middle

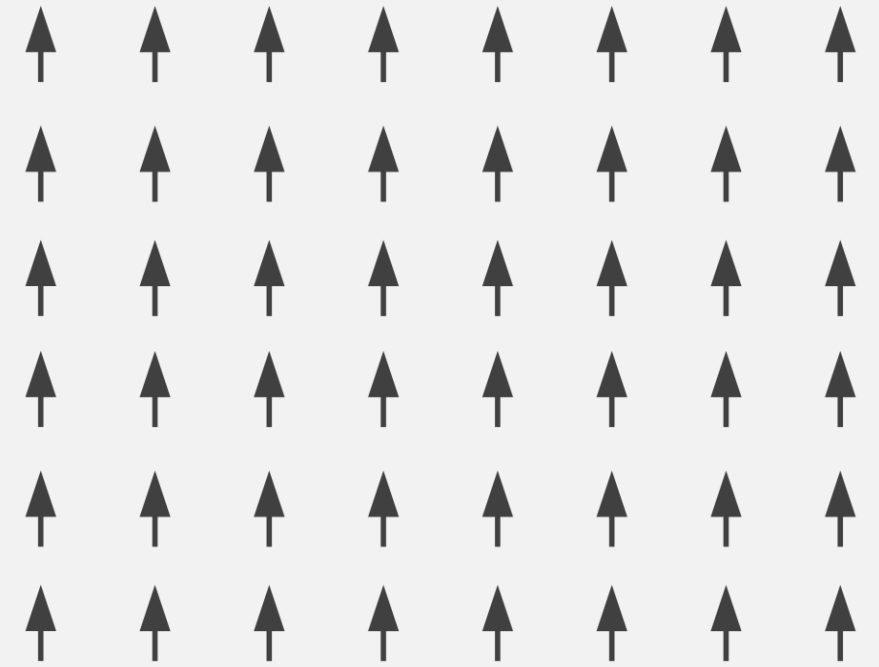
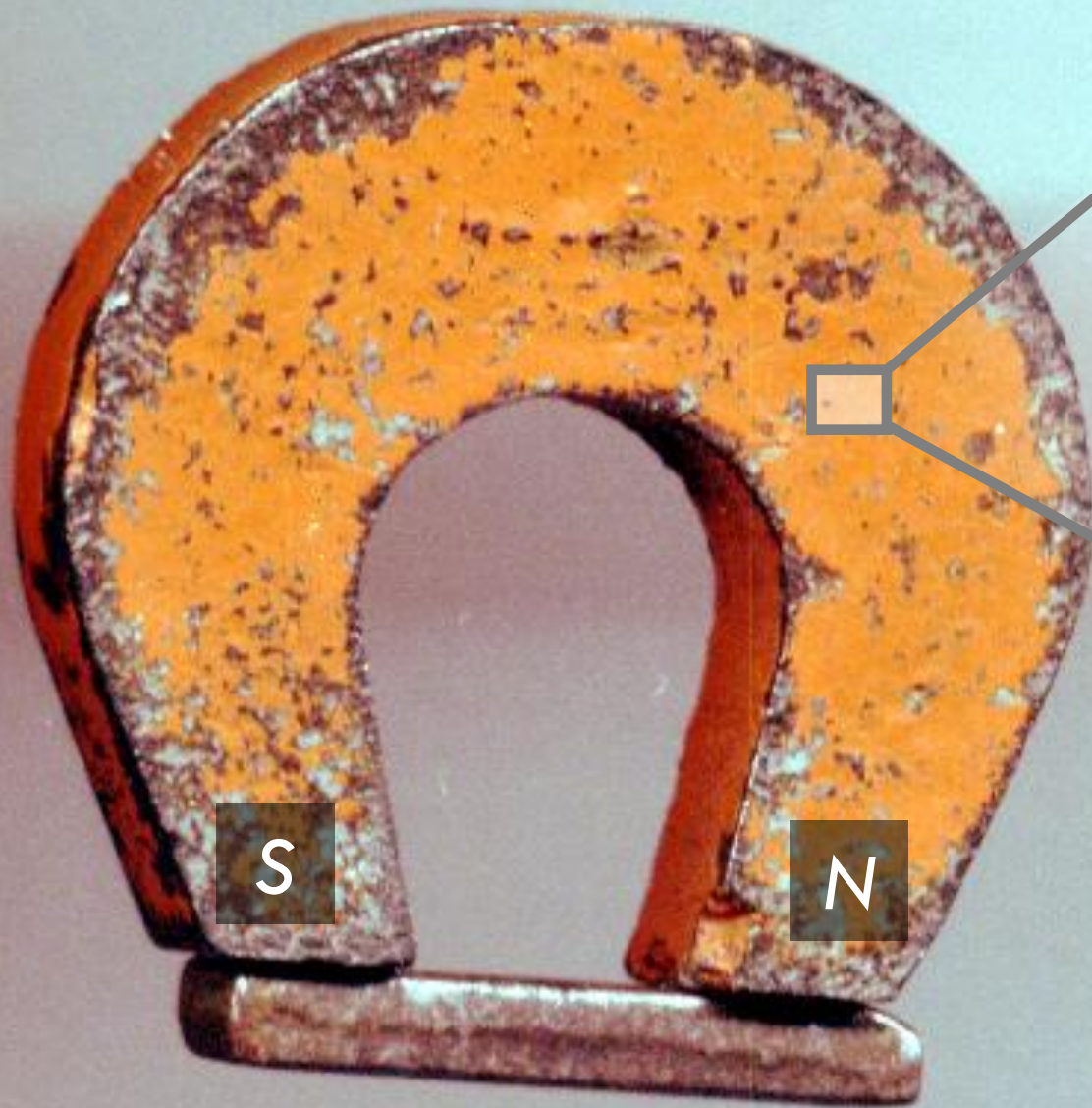
quartzpage.de

Locations of Si & O





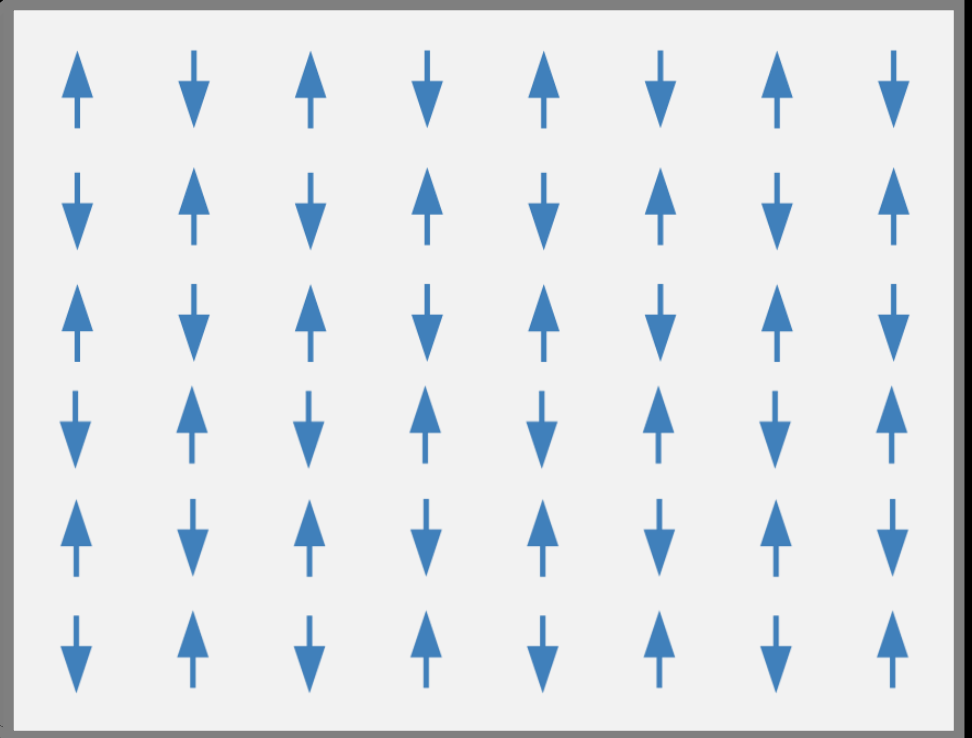
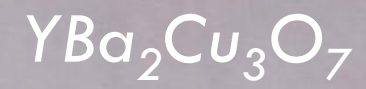
Repeating “unit” of
the crystal



*Atomic spins are **aligned***

*Regular arrangement of
magnetic dipoles*

Ferromagnet



*Atoms alternate between
aligned and anti-aligned*

Still regular arrangement!

High-Temperature Superconductor

So how do define “order”?

Local Observable? Physical property in one place.

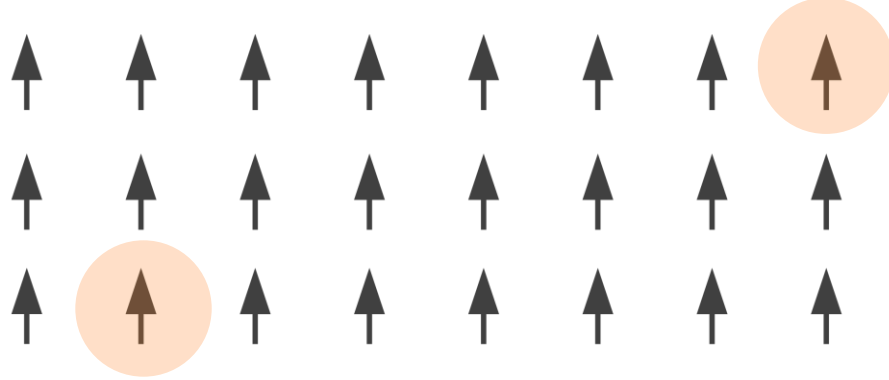
E.g. Location of an atom, direction of a spin, ...

Definition: A system has (long-range) order if some **local observable property** is **correlated** even at very **distant locations**

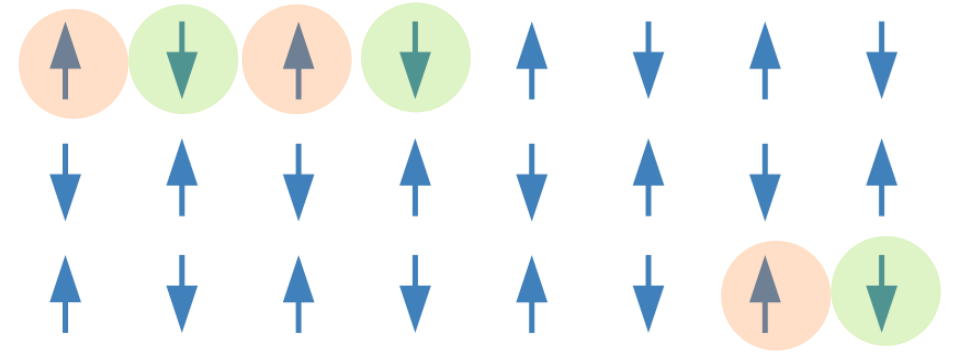
Distant? Many, many atomic spacings apart

Correlated? We can (mostly) determine the value of one from the other

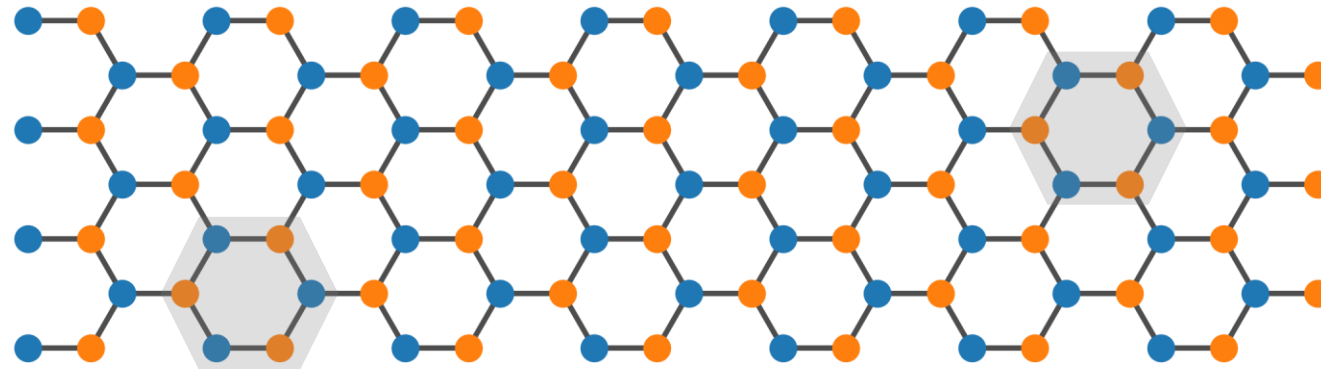
If you know this direction ...



... you know this one too



(Same idea, just alternate direction)



*Once you fix the first few atoms,
the **pattern** fixes all the rest*

Why do we find order?

- At low temperatures, systems tend to states with **as little energy as possible**

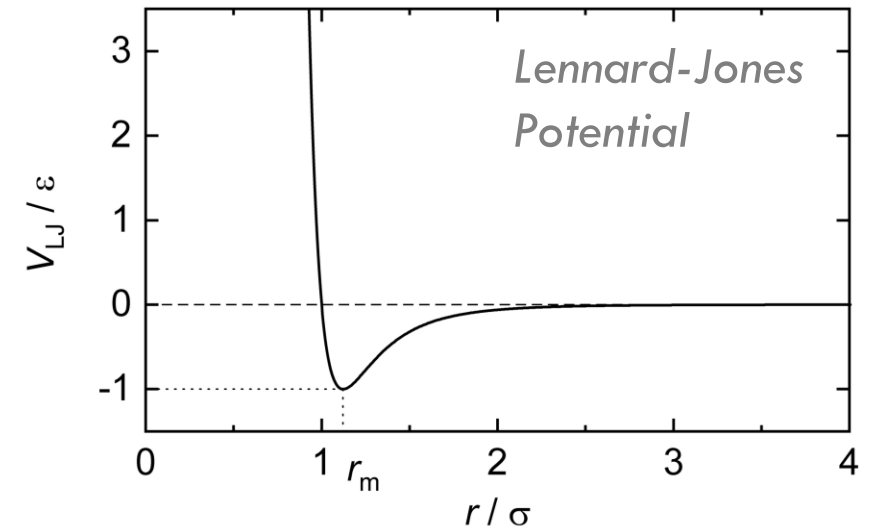
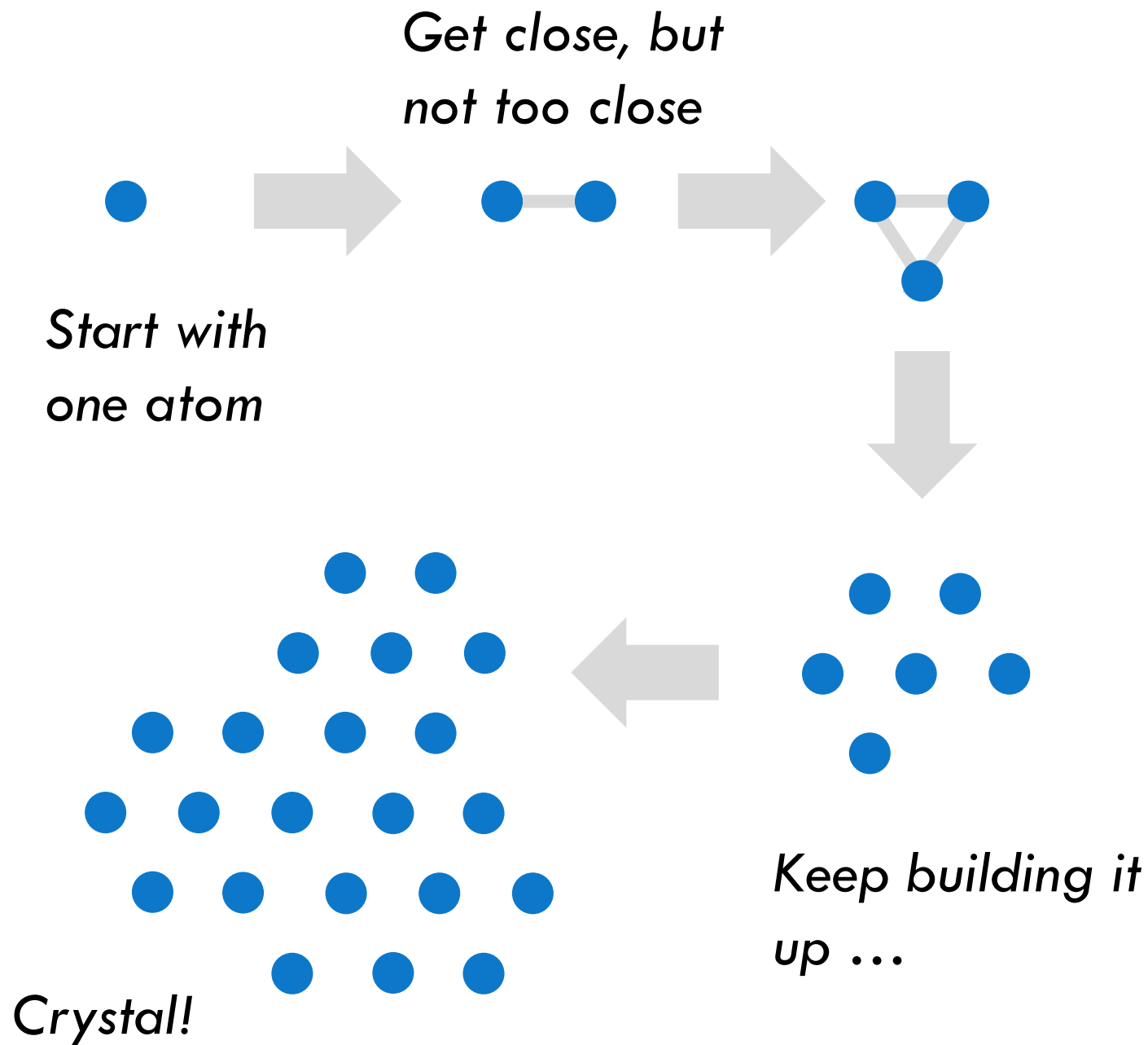
Cold removes all the energy it can from system

- Interactions between atoms typically **highly symmetric**

Laws of Nature have high symmetry!

- Only* a **handful** of way to get the lowest possible energy

** Not always true: “Frustration”*



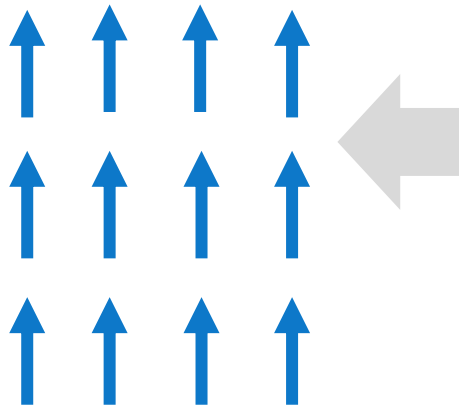
*All things are made of atoms—
little particles that move around in
perpetual motion, **attracting each
other when they are a little
distance apart, but repelling upon
being squeezed into one another.***

*In that one sentence ... there is an
enormous amount of information
about the world. - **R. P. Feynman***

Align its neighbour

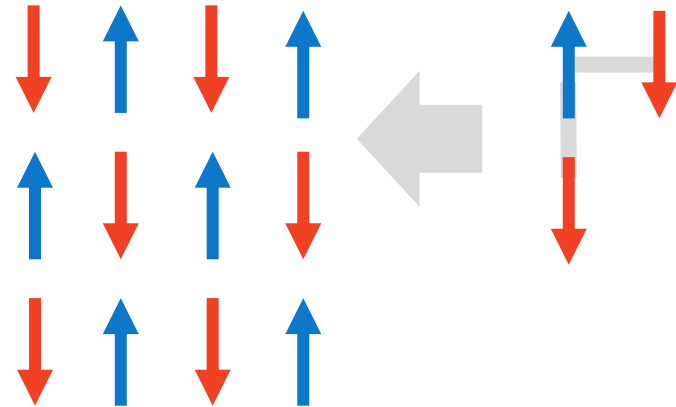


*Start with one
magnetic moment*



*Build up full
lattice*

***Anti-align** its neighbour*



*Build up alternating
pattern*

Three questions:

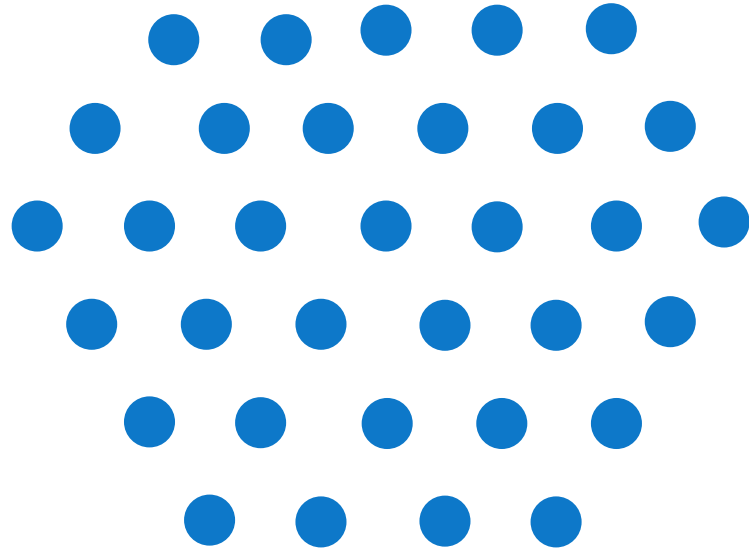
1. What is **order**?

2. What is **disorder**?

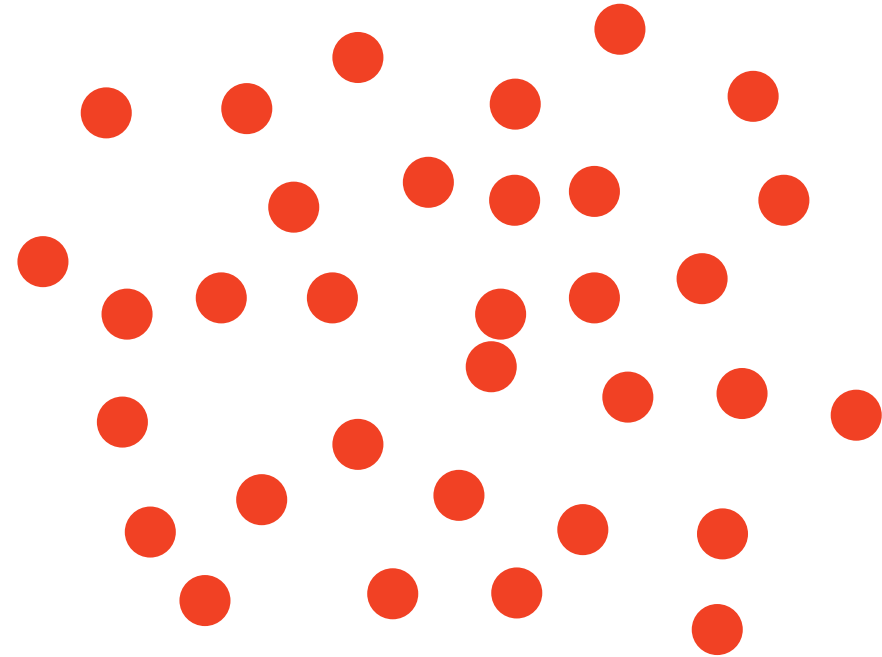
3. How can **disorder** lead to **order**?

Can't we just say it is what order *isn't*?

- If order is **correlation at long distances** disorder a **lack thereof**?

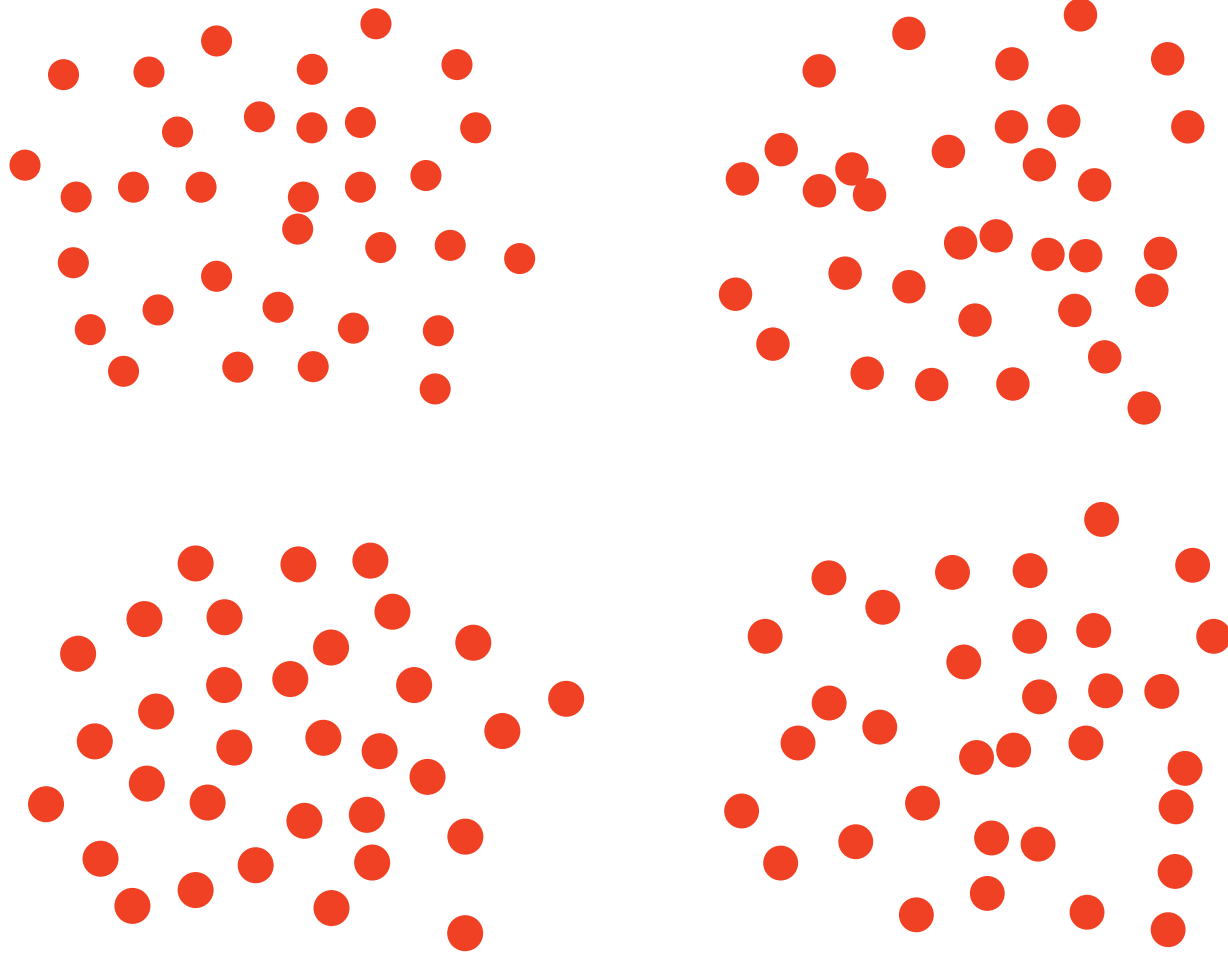


Order



Disorder?

Lots of disordered states “look” the same

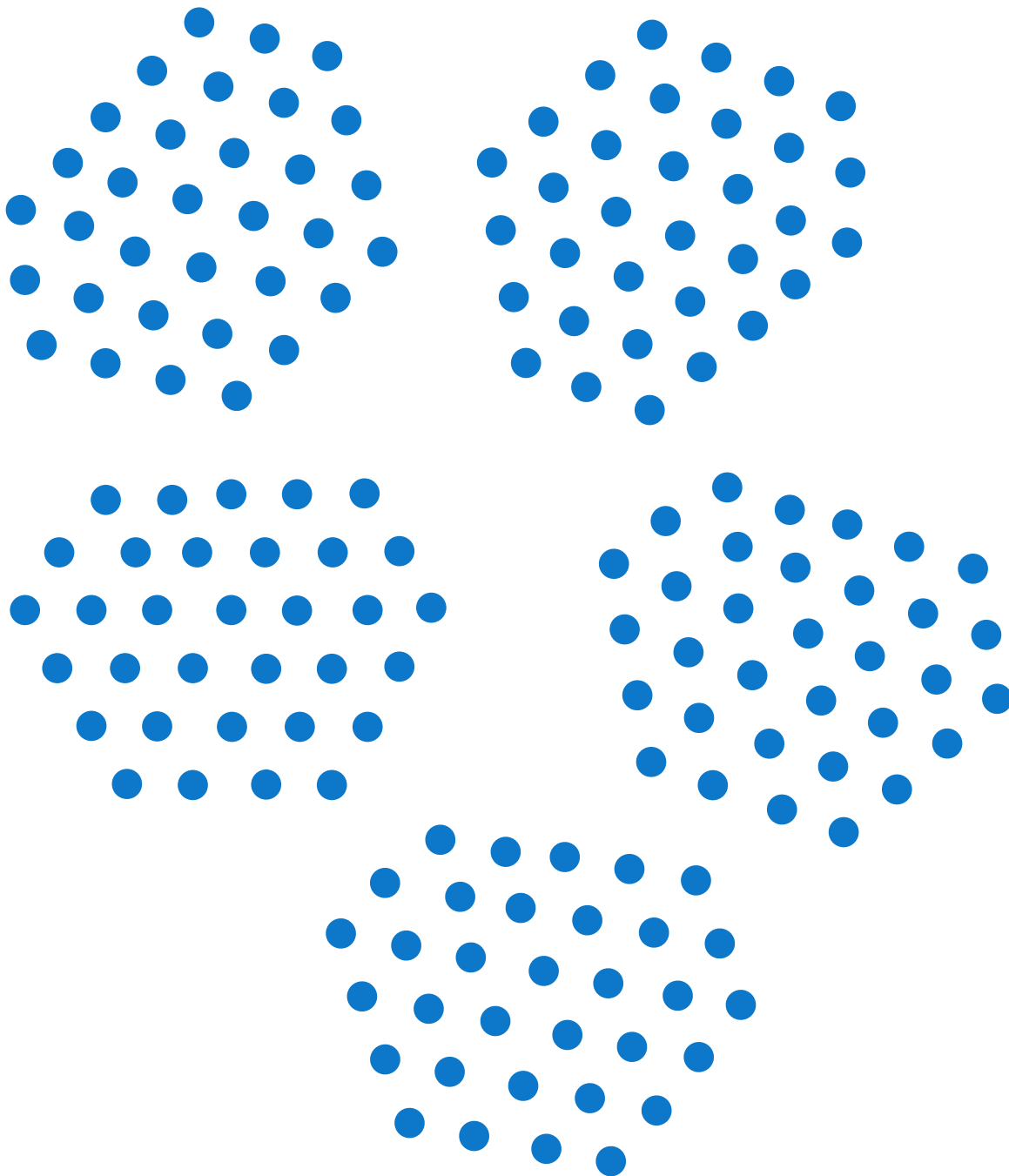


- **Qualitatively** these different disordered states are distinguishable *by eye*
- Lots of atoms have been moved ...
- ... but whatever **macroscopic** properties we're seeing don't look different!

Ordered states *aren't* like this

- All we can do is a few simple changes:
 - **Translation:** Shift the atoms all together
 - **Rotation:** Rotate the atoms all together

*How do we make this **precise**?*

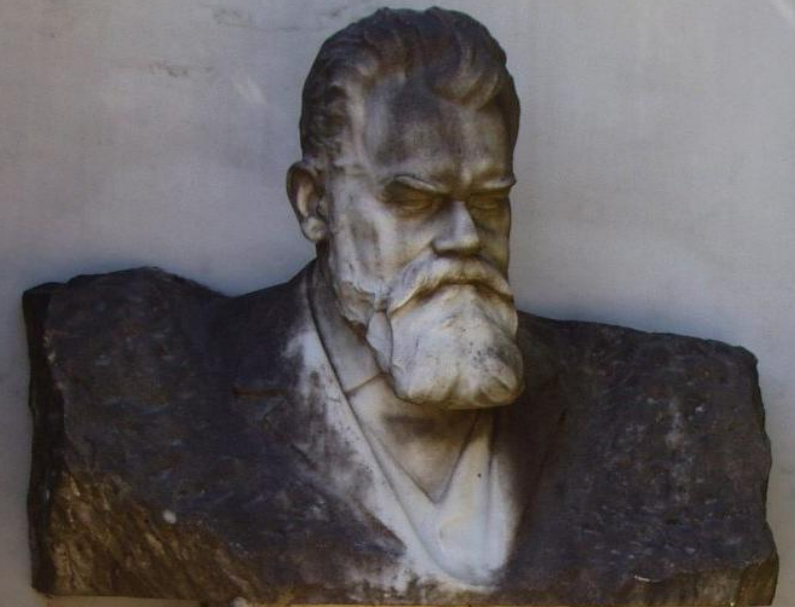


$$S = k \cdot \log W$$

$$S = k \cdot \log W$$

The general struggle for existence of animate beings is not a struggle for raw materials, these for organisms are air water & soil, all abundantly available, nor for energy which exists in plenty in the sun and any hot body in the form of heat, but rather a struggle for entropy, which becomes available through the transition of energy from the hot sun to the cold earth.

- Ludwig Boltzmann



LUDWIG
BOLTZMANN
1844 - 1906

DR. PHIL. PAUL
BOLTZMANN
GEB. CHIARI
1891 - 1977
ARTHUR

Entropy

Why a logarithm? W gets really, really big

Definition: The **entropy** of a system is:

$$\text{Entropy } S = k_B \log W$$

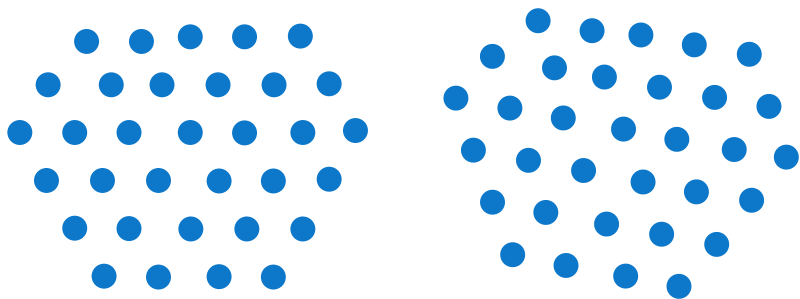
Number of states consistent with our macroscopic description

Boltzmann's constant (just sets units)

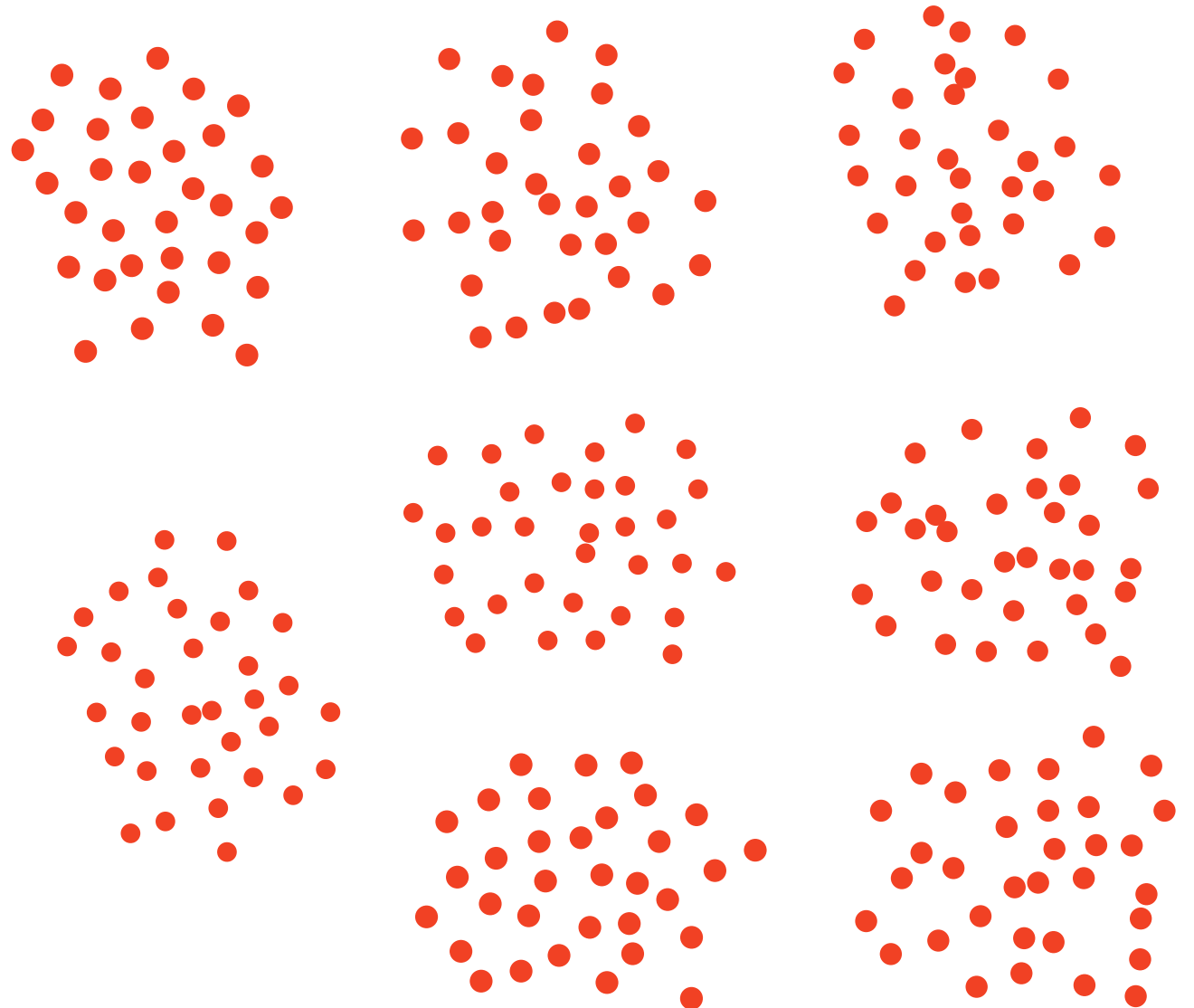
- What do we mean by **macroscopic description**?
Almost anything involving the state of all the atoms at once
- **Examples:** Total Energy, Number of Particles, Volume, Pressure, Magnetization, ...

For the gas?

- Simple macroscopic variable: **the total energy**
- Lower when atoms near *preferred* distance



Atoms *minimizing* energy



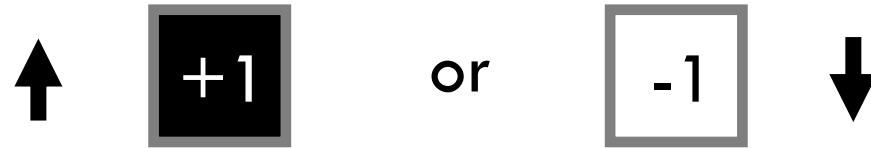
Atoms too far & too close, energy *higher*

Simple Example: Ising Model



$$M = 0$$

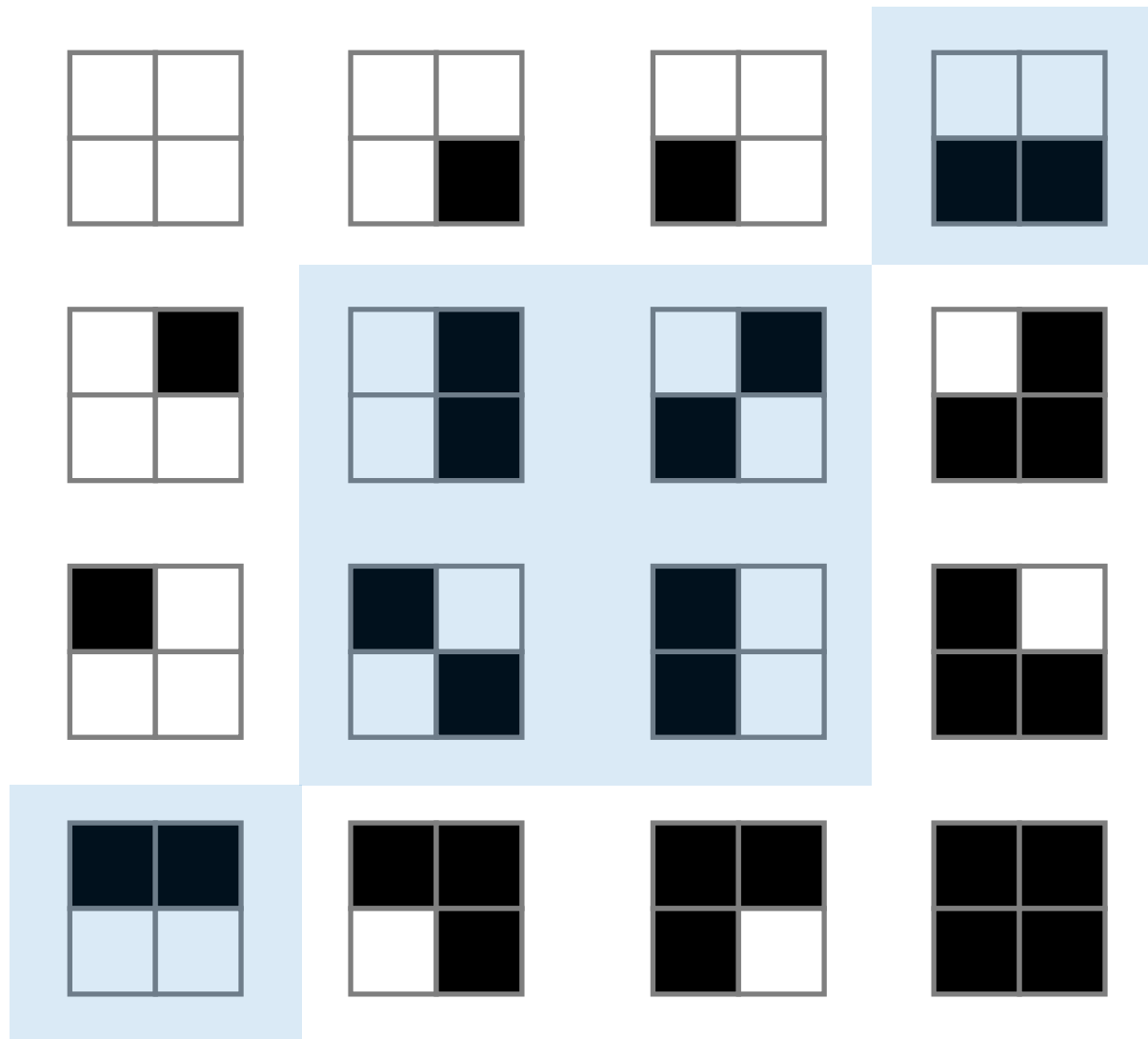
- Array of variables, each taking one of two values



- This could be **spin**, atom type, ...
- **Macroscopic state?**

$$M = (\text{Sum over all variables})$$

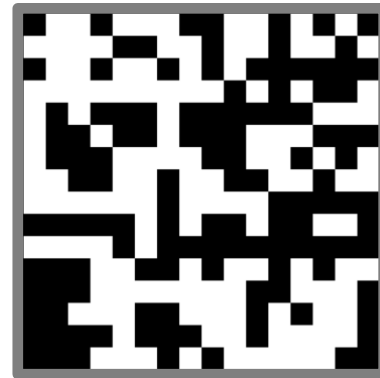
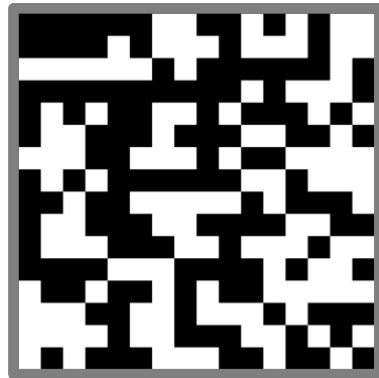
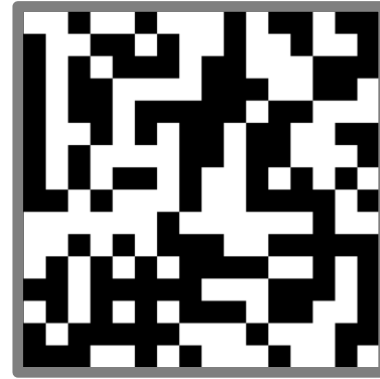
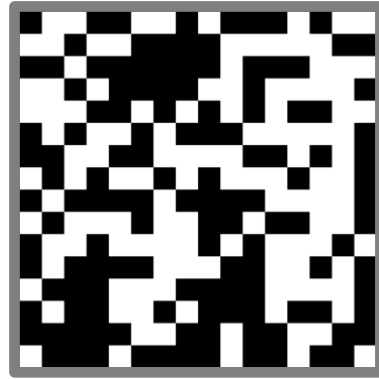
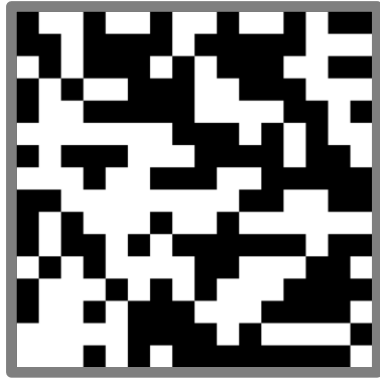
$$M=0$$



$$N=4$$

- Look at small case: **just four variables**
- Say we are only talking about states where **$M=0$**
- **Fewer** states with $M=+2, -2$ or $M=+4, -4$
- Discrepancy *grows* with N

Lots and lots of states with $M = 0 \dots$



High entropy, look “disordered” in usual sense

How many states?

$N=4096$

- When number of variables is large, **exponentially many** with $M=0$, roughly $\sim 2^N$

$$S \sim N \log 2$$

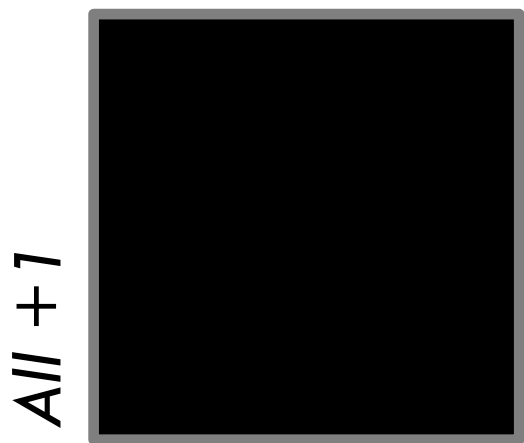
- Exponentials grow **fast**
 - $2^{16} \sim 65,536$
 - $2^{32} \sim 4,294,967,296$
 - $2^{64} \sim 18,446,744,073,709,551,616$



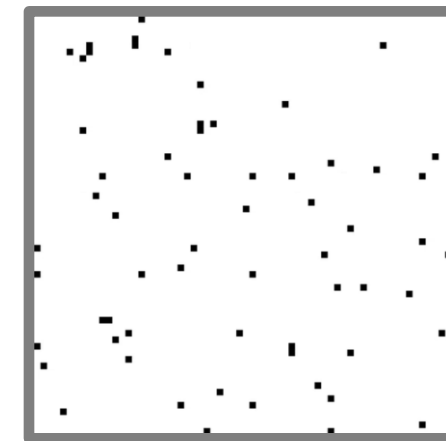
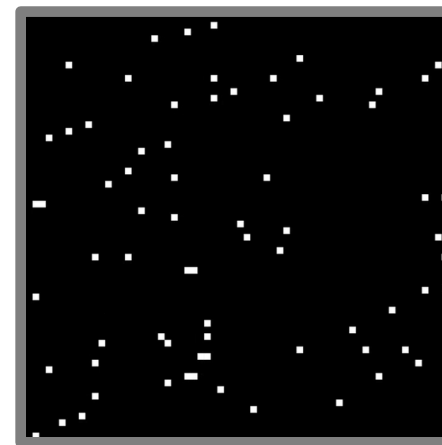
Ordered States?

- **Two** possibilities: **All +1** or **All -1**
- Entropy?

$$S \sim \log(2)$$



Near maximum M ?



- Variety of states much lower even **near** all +1 or all -1
- Entropy *shrinks* with larger M

Why do we care about this?

Fundamental principle of statistical physics:

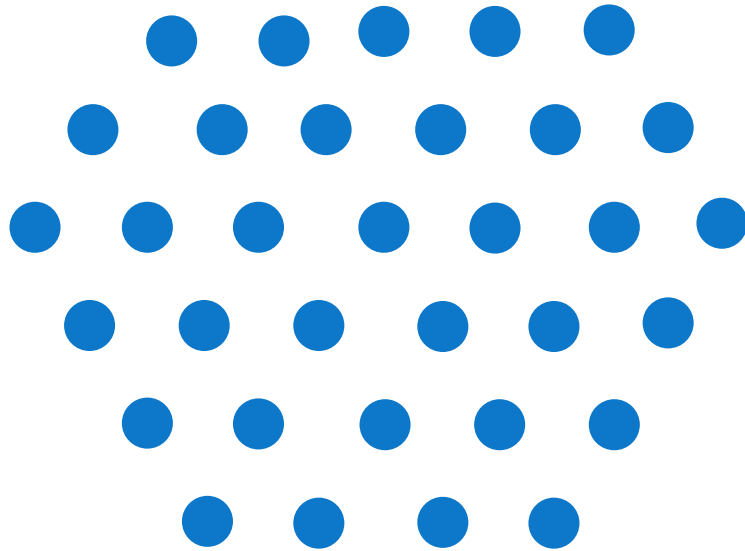
*All else being equal, a system is most likely to be found in (accessible) states that **maximize its entropy***

Consistent with our macroscopic description

Principle of “Ignorance”: Assume all (accessible) states are equally likely. You’ll end up in high entropy states just because **there are more of them.**

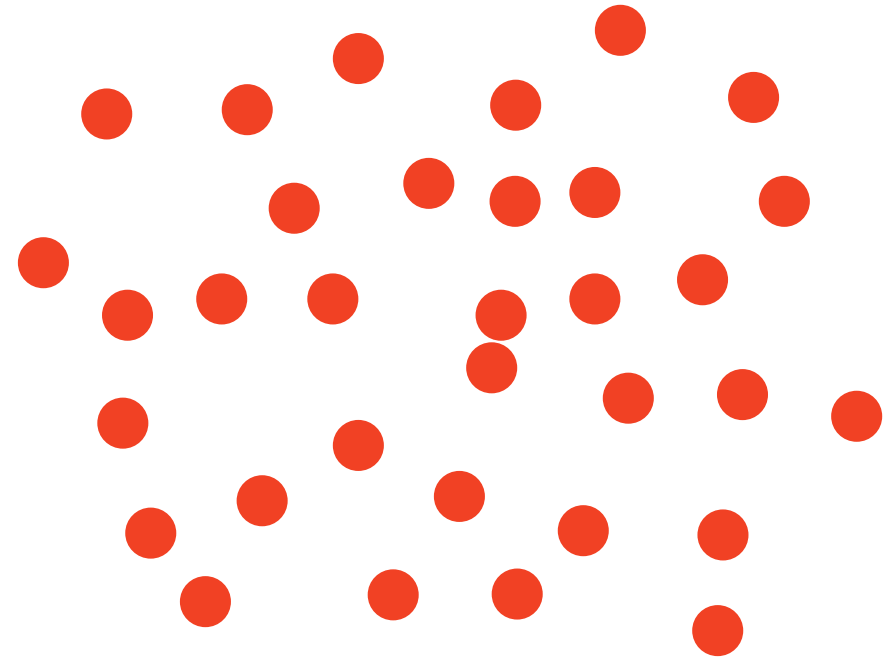
Competition between entropy and energy

Low-entropy
Low energy



Favoured at low temperature

High entropy
High energy



Favoured at high temperature

Three questions:

1. What is **order**?

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3. How can **disorder** lead to **order**?

Three questions:

1. What is order?

2. What is disorder?

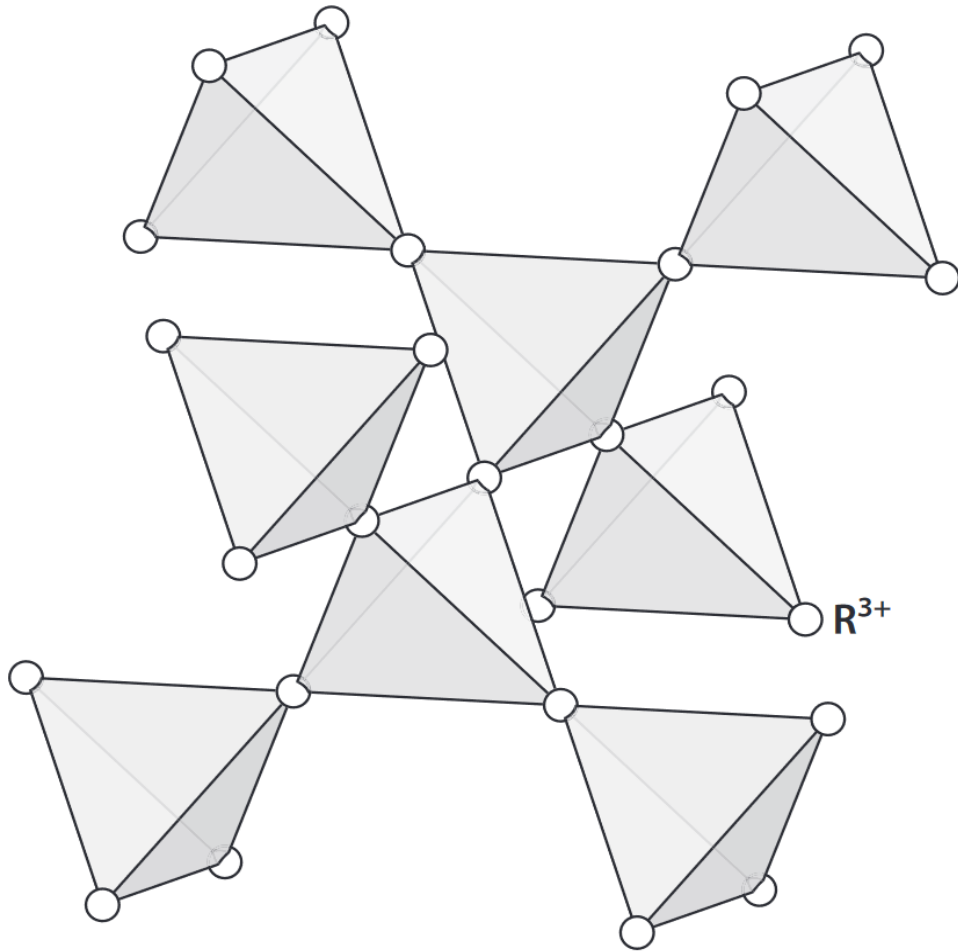
entropy

3. How can ~~disorder~~ lead to order?

Answer:

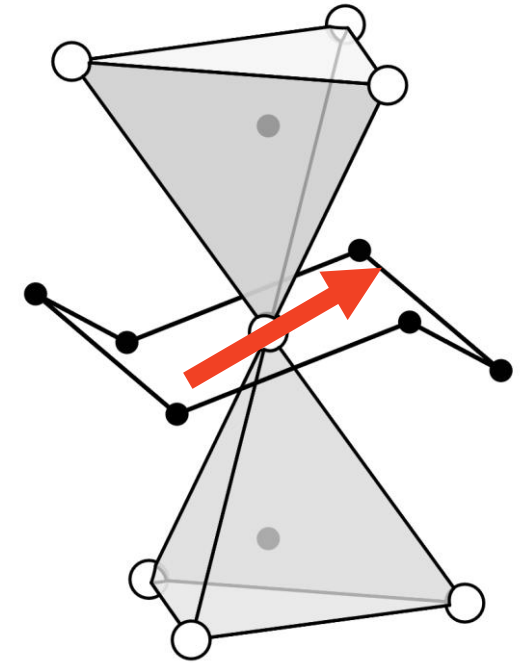
If the energies are all the same, it is left to the **entropy** to decide which state is preferred

Erbium Titanate



Chemical formula: $\text{Er}_2\text{Ti}_2\text{O}_7$

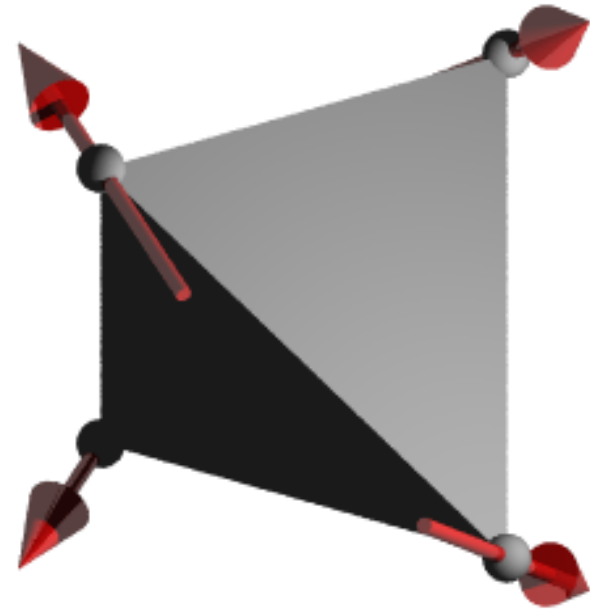
- Magnetic ion is a trivalent **Erbium** (rare-earth)
- Forms *three-dimensional pyrochlore lattice*
 - Network of corner-sharing tetrahedra



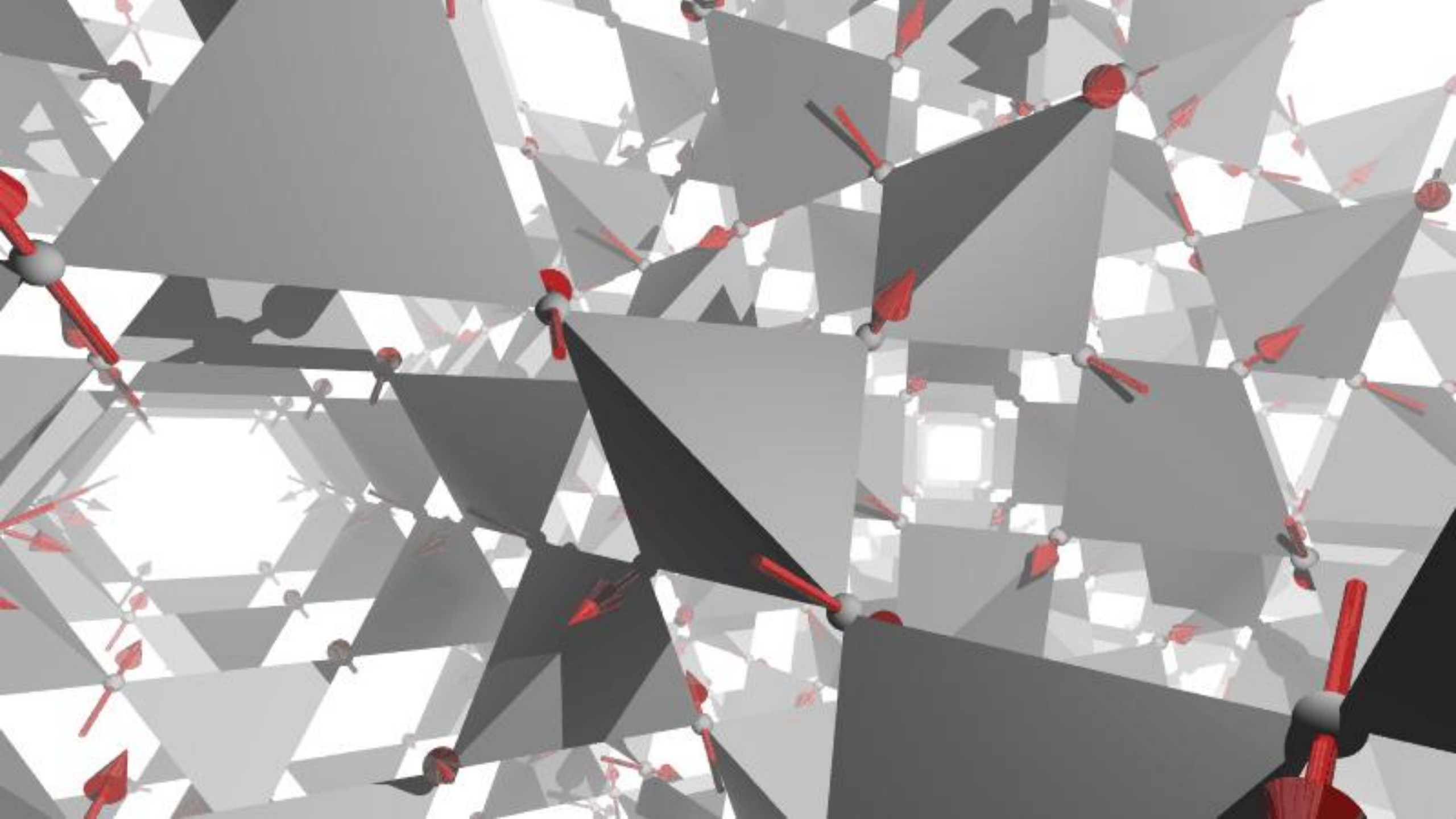
Each atom has a *spin*, pointing in some direction

Minimum energy states?

- Energy of different arrangements of spins is **complicated**
- Minimum has each tetrahedron with *same* spin configuration
- Spins can be **rotated together in their planes** at no (classical) cost



State with minimum energy isn't unique

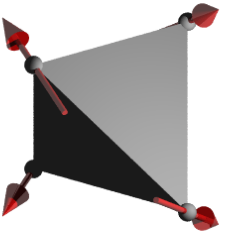
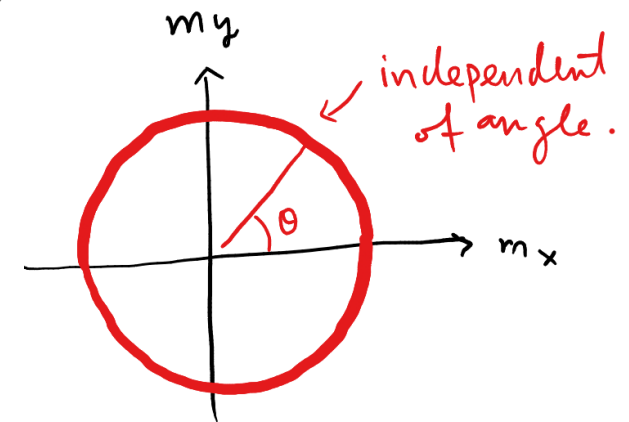
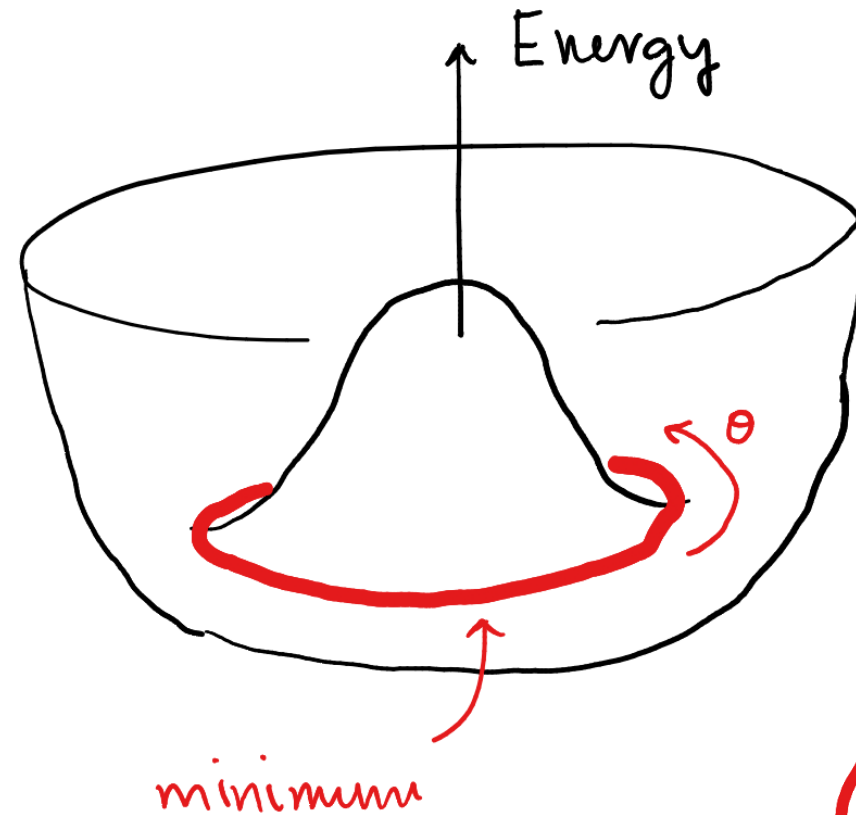


When energy is not enough ...

- We have a system where there are **many states with the same energy** ...

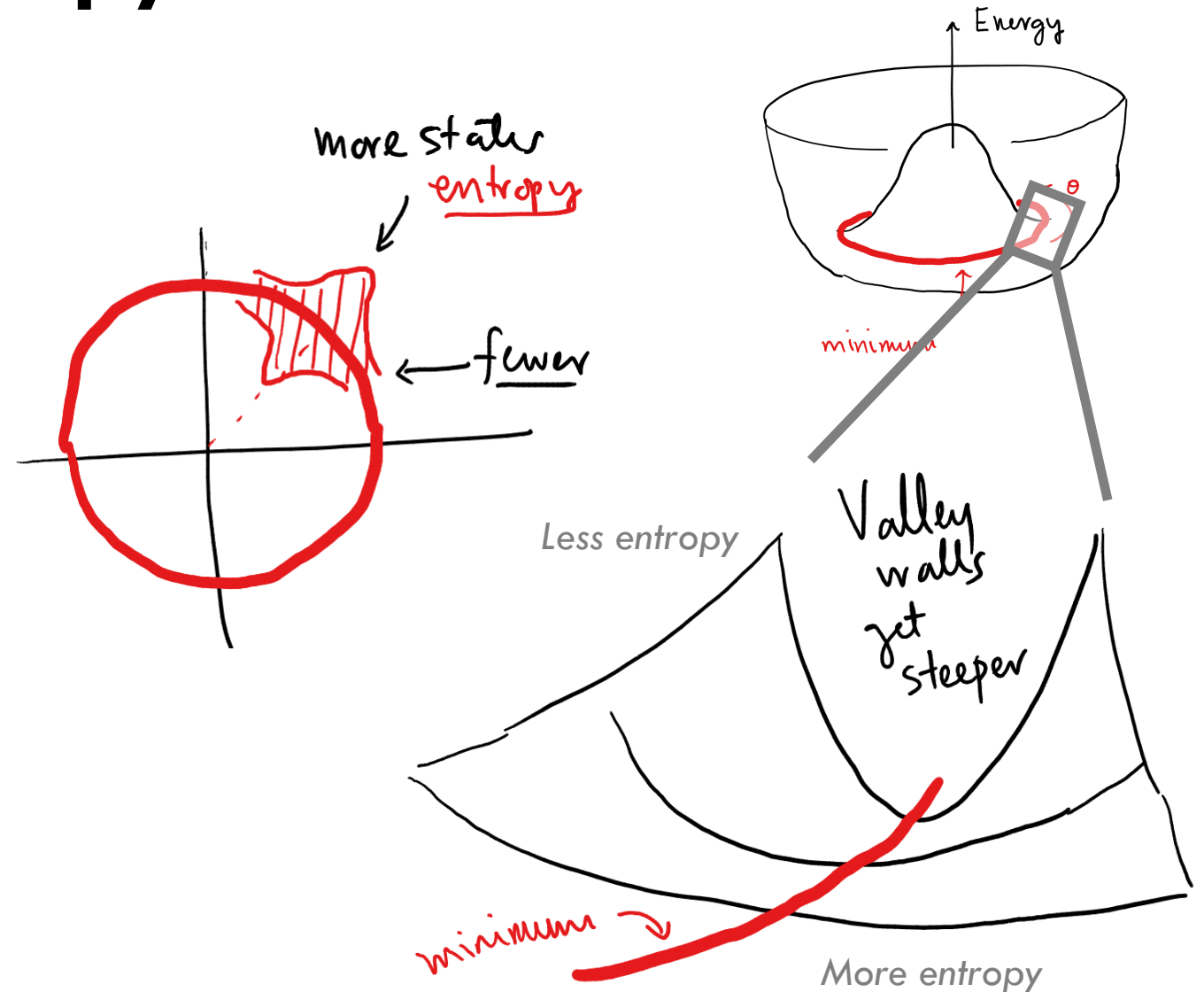
Which state does it pick?

Answer: None!



What about the entropy?

- Even though the energy is the same, the **entropy might not have to be**
- Look at how many states there are **just off the minimum**
- Regions with **higher entropies** are chosen!



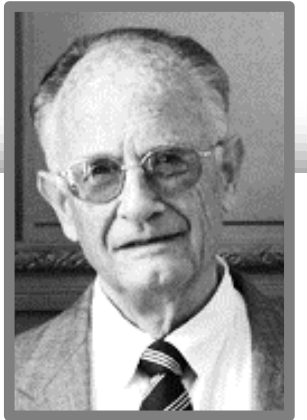
Order *by* Disorder

So what happened?

- Without entropy: **No order**
- With entropy: **Order**

- Not limited to only *thermal* disorder

Any kind of fluctuations can serve same purpose!



Jacques Villain
(1934-2022)

J. Physique **41** (1980) 1263-1272

Classification
Physics Abstracts
75.10H

Order as an effect of disorder

J. Villain (*), R. Bidaux, J.-P. Carton and R. Conte

DPh-G/P \bar{S} RM, CEN de Saclay, B.P. N $^{\circ}$ 2, 91190 Gif-s/Yvette, France

(Reçu le 9 avril 1980, révisé le 3 juillet, accepté le 11 juillet 1980)

Résumé. — On considère un modèle d'Ising frustré généralisé sur un réseau bidimensionnel magnétique à température nulle mais ferromagnétique pourvu que $0 < T < T_c$. La dilution sur ce système, et l'on montre que l'ordre à longue distance est rétabli dans certaines conditions de concentration, température et interactions qui sont discutées en comparaison avec le cas usuel.

Abstract. — A generalized frustrated Ising model on a two-dimensional lattice is considered magnetic at zero temperature but ferromagnetic provided $0 < T < T_c$. The effect of dilution on this system, and long range order is shown to be restored in the dilute model under certain conditions of concentration, temperature and interactions which are discussed in comparison with the usual case.

Quantum Fluctuations?

Heisenberg's Uncertainty Principle

$$\Delta x \Delta p \geq \frac{\hbar}{2}$$

Fluctuation in momentum

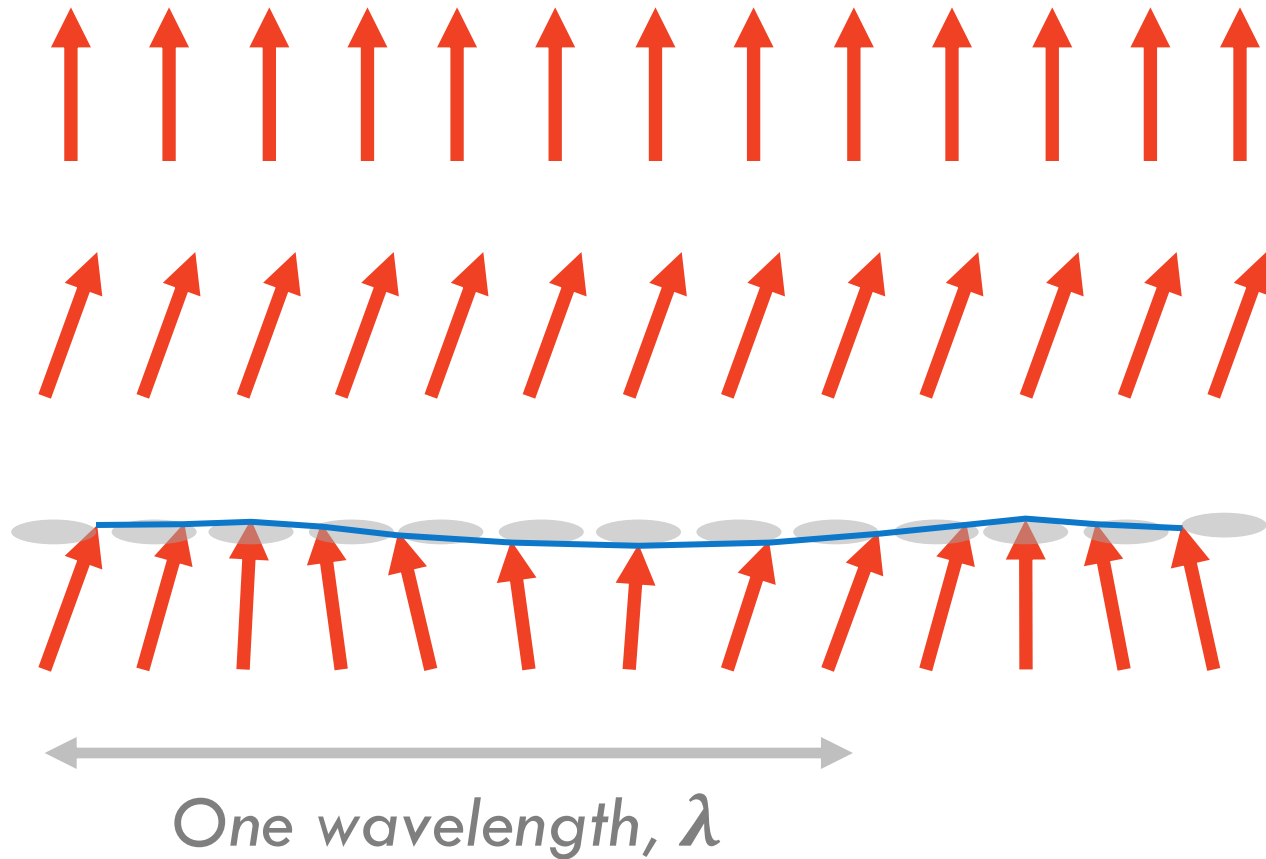
Fluctuation in position

Planck's Constant (reduced)

- Order by *thermal* disorder needs finite temperature
- Even at **zero temperature** a system can fluctuate due to **quantum mechanics**
- Preference? ***More fluctuations!***

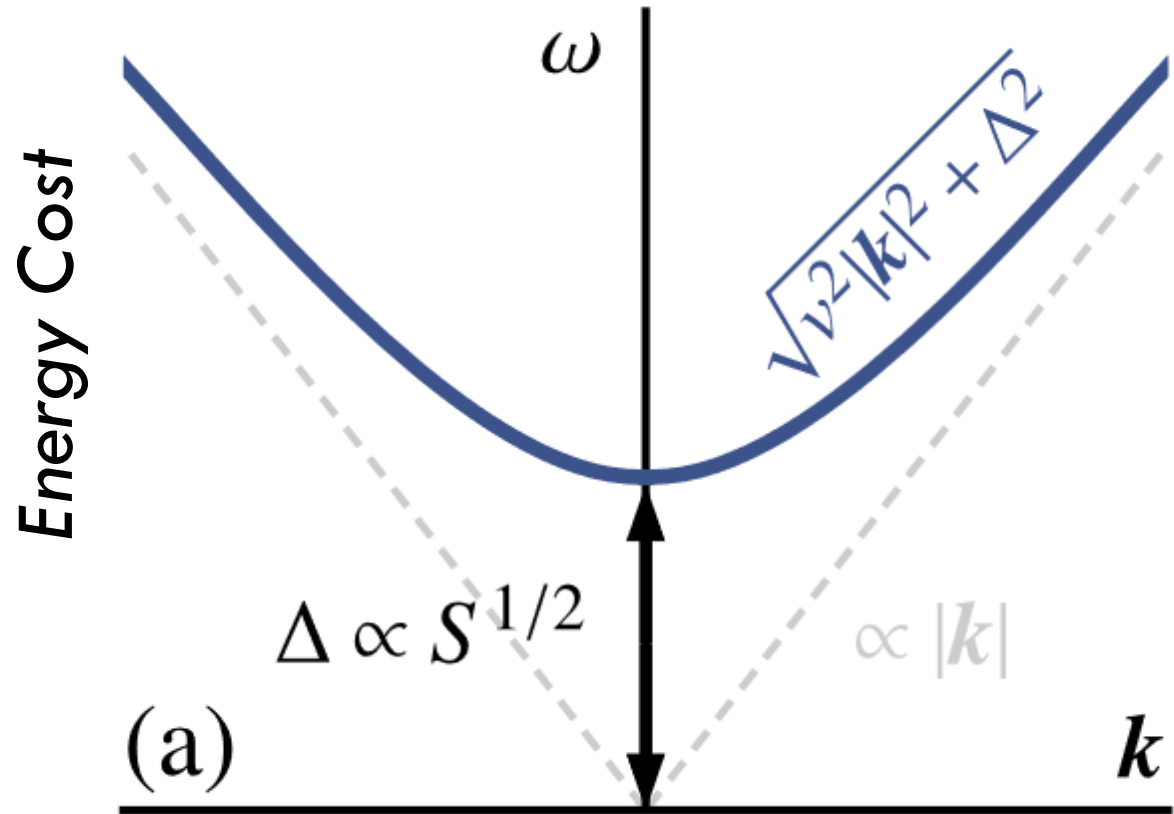
Cannot have **no fluctuations**; one of x or p must be fluctuating

How do we detect it?



- *Without fluctuations* we have states at the **same energy**
- No cost to change all at once
- **Small** cost for change over long distances – **“Goldstone mode”**

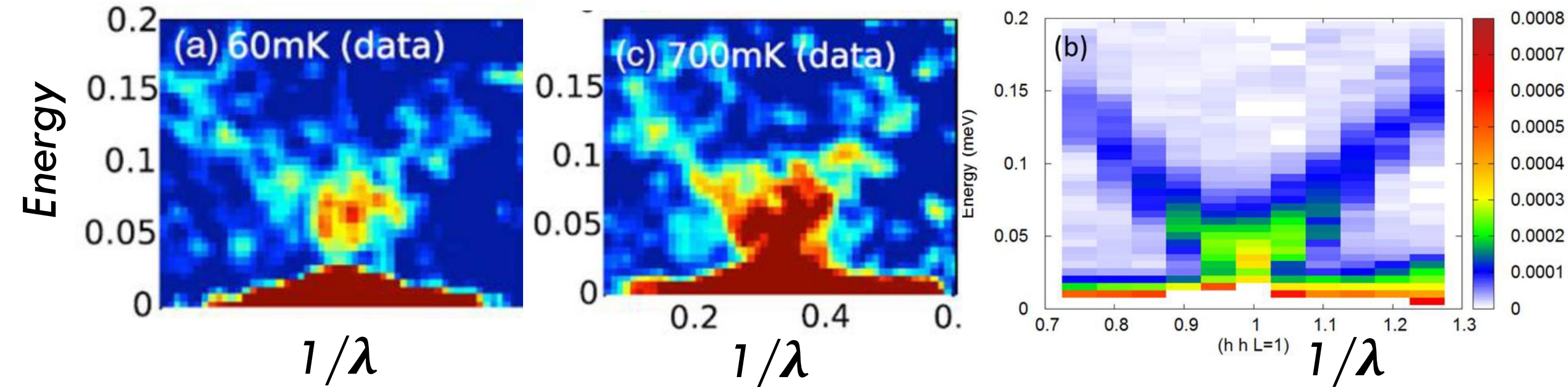
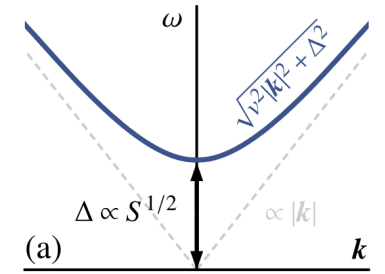
Pseudo-Goldstone Modes



Wave-number, $1/\lambda$, of deformation

- Well defined relationship between **wavelength** or **wavenumber** and **energy cost**
- Called a “**dispersion relation**”
- *Experimentally observable*

Measurement of Gap



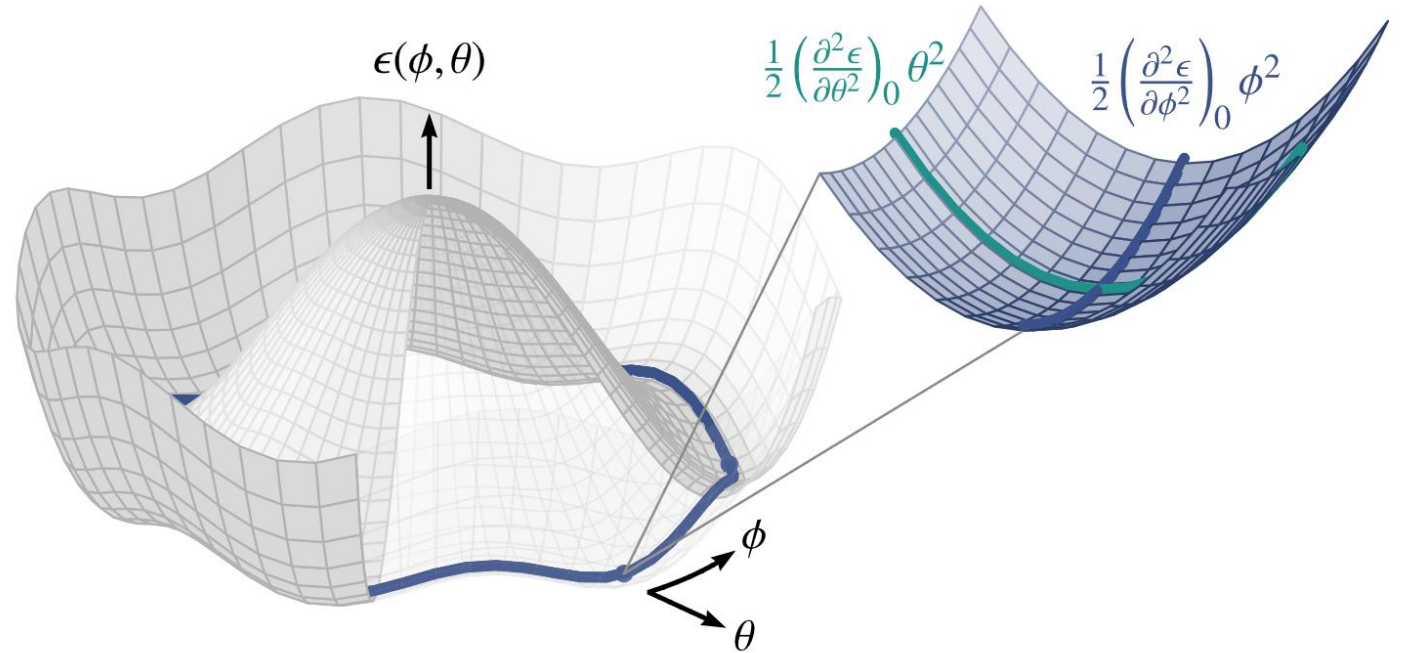
- There is a **gap**!
 - Measured independently by two experimental groups
- Is it what we expect from quantum fluctuations?

Gap is roughly
 $\sim 0.04 - 0.05 \text{ meV}$

How to compute the gap?

- Can relate it **directly to curvatures of “fluctuation free energy”**

(Proof somewhat involved)



$$\Delta = \frac{1}{S} \sqrt{\left(\frac{\partial^2 \epsilon}{\partial \theta^2}\right)_0 \left(\frac{\partial^2 \epsilon}{\partial \phi^2}\right)_0 - \left(\frac{\partial^2 \epsilon}{\partial \theta \partial \phi}\right)_0^2}$$

Pseudo-
Goldstone Gap

Derivatives of “fluctuation
free energy”

This formula works
for thermal ObD **and**
quantum ObD

Er₂Ti₂O₇
[61,67,68]

Savary *et al.*
[61]

I

31.1 μeV

43 – 53 μeV
[82,83]

TABLE I. Calculations showing the equality of the pseudo-Goldstone gap, Δ , computed from nonlinear spin-wave theory [Eqs. (5) and (6)] and then independently from the curvatures of the classical and quantum zero-point energies [Eq. (1)]. For each model, the lattice, the exchange regime, the type of pseudo-Goldstone mode, and several choice of parameters are listed. When available, additional theoretical or experimental estimates of the pseudo-Goldstone gap are shown.

Model or material	Parameters	Type	Δ	$[(\partial^2\epsilon/\partial\theta^2)]_0$	$[(\partial^2\epsilon/\partial\phi^2)]_0$	$S^{-1}\sqrt{[(\partial^2\epsilon/\partial\theta^2)]_0[(\partial^2\epsilon/\partial\phi^2)]_0}$	$S = \frac{1}{2}/\text{Exp.}$
Heisenberg-compass (Square, Ferromagnet)	$ K / J \ll 1$	I	$0.52S^{\frac{1}{2}} K ^{\frac{1}{2}}/ J ^{\frac{1}{2}}$	$2 K S^2$	$0.137K^2S/ J $	$0.52S^{\frac{1}{2}} K ^{\frac{1}{2}}/ J ^{\frac{1}{2}}$	
	$K/ J = -0.5$	I	$0.17 J S^{\frac{1}{2}}$	$ J S^2$	$0.0286 J S$	$0.17 J S^{\frac{1}{2}}$	
Heisenberg-compass [16] (Cubic, Ferromagnet)	$ K / J \ll 1$	II	$0.093K^2/ J $	$0.093K^2S/ J $	$0.093K^2S/ J $	$0.093K^2/ J $	
	$K/ J = +0.5$	II	$0.030 J $	$0.030 J S$	$0.030 J S$	$0.030 J $	
	$K/ J = -0.5$	II	$0.024 J $	$0.024 J S$	$0.024 J S$	$0.024 J $	
Heisenberg-Kitaev [60] (Honeycomb, Ferromagnet)	$ K \ll J $	II	$0.0897K^2/ J $	$0.0897K^2S/ J $	$0.0897K^2S/ J $	$0.0897K^2/ J $	
	$K/ J = -2.0$	II	$0.208 J $	$0.208 J S$	$0.208 J S$	$0.208 J $	
	$K/ J = -0.65$	II	$0.03 J $	$0.0300 J S$	$0.0300 J S$	$0.0300 J $	$\sim 0.05 J $ [80]
Heisenberg-Kitaev [60] (Honeycomb, Néel)	$ K \ll J$	I + I	$0.83 K S^{\frac{1}{2}}$	$2(3J + K)S^2$	$0.115K^2S/J$	$0.83 K S^{\frac{1}{2}}$	
	$K/J = +2.0$	I + I	$1.66JS^{\frac{1}{2}}$	$10JS^2$	$0.274JS$	$1.66JS^{\frac{1}{2}}$	
	$K/J = -0.5$	I + I	$0.434JS^{\frac{1}{2}}$	$5JS^2$	$0.038JS$	$0.434JS^{\frac{1}{2}}$	
Heisenberg- Γ [62] (Honeycomb, Ferromagnet)	$\Gamma \ll J $	I	$0.29\Gamma^2/ J S^{\frac{1}{2}}$	$3\Gamma S^2$	$0.028\Gamma^3S/ J ^2$	$0.29\Gamma^2/ J S^{\frac{1}{2}}$	
	$\Gamma/ J = +0.5$	I	$0.081 J S^{\frac{1}{2}}$	$1.5 J S^2$	$0.00437 J S$	$0.081 J S^{\frac{1}{2}}$	
	$\Gamma/ J = +1.0$	I	$0.355 J S^{\frac{1}{2}}$	$3 J S^2$	$0.042 J S$	$0.355 J S^{\frac{1}{2}}$	
$J_1 - J_2$ [7,11,13] (Square, stripe)	$J_1/J_2 \ll 1$	I + I	$1.44J_1S^{\frac{1}{2}}$	$4(2J_2 - J_1)S^2$	$0.2604J_1^2S/J_2$	$1.44J_1S^{\frac{1}{2}}$	
	$J_1/J_2 = 0.5$	I + I	$0.63J_2S^{\frac{1}{2}}$	$6J_2S^2$	$0.0668J_2S$	$0.63J_2S^{\frac{1}{2}}$	$0.61J_2S^{\frac{1}{2}}$ [81]
	$J_1/J_2 = 1$	I + I	$1.08J_2S^{\frac{1}{2}}$	$4J_2S^2$	$0.294J_2S$	$1.08J_2S^{\frac{1}{2}}$	$0.96J_2S^{\frac{1}{2}}$ [81]
$J_1 - J_2$ [13,14] (Triangular, stripe)	$J_2/J_1 = 0.25$	II + II	$0.53J_1$	$0.53J_1S$	$0.53J_1S$	$0.53J_1$	
	$J_2/J_1 = 0.5$	II + II	$0.45J_1$	$0.45J_1S$	$0.45J_1S$	$0.45J_1$	
	$J_2/J_1 = 0.75$	II + II	$0.58J_1$	$0.58J_1S$	$0.58J_1S$	$0.58J_1$	
Er ₂ Ti ₂ O ₇ [61,67,68]	Savary <i>et al.</i> [61]	I	31.1 μeV	157.5 μeV	1.536 μeV	31.1 μeV	43 – 53 μeV [82,83]
Ca ₃ Cr ₂ Ge ₃ O ₁₂ [8,65]	Brückner <i>et al.</i> [65]	I + I	262 μeV	4 μeV	107.5 μeV	262 μeV	156 μeV [83]

Theoretical
Value

Experimental
Value

- Theoretical value is of **right order of magnitude**
- **Quantitative?** Off by 30%-50%
- **How to improve?** Better model/energy, better experiment
- ...

Summary & Conclusions

- **Order and disorder are fundamental in physics**
 - Long-range order characterizes many phases of matter
 - Entropy counts the number of states compatible with macroscopic properties
 - At finite temperature, entropy and energy compete
- **Lots to explore!**
 - Better understanding of the variety of forms of order by disorder in **real materials**
 - **Frustration:** States with **low-energy, but high entropy**.
What happens when we include fluctuations?

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Thank you
for your
attention